A Dash Of Style The Art And Mastery Punctuation Noah Lukeman

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web the carbon hydrogen chains in vulcanized rubber are held together by two sulfur atoms that form covalent bonds between the chains these covalent bonds are commonly called disulfide bridges explore other molecules that have such disulfide bridges present your findings to the class

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web because of the fact that polar
covalent bonds have higher melting and
boiling points why do dipole forces
have an impact on the melting and
boiling points of substances this is
because the dipole forces are an added
force of attraction that slows the
molecules down making it take more
energy to melt or boil them